Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

S. Sathiya Moorthi,^a K. Chinnakali,^b* S. Nanjundan,^c R. Balaji^d and Hoong-Kun Fun^e*

^aDepartment of Physics, Sri Muthukumaran Institute of Technology, Chikkarayapuram, Chennai 600 069, India, ^bDepartment of Physics, Anna University, Chennai 600 025, India, ^cDepartment of Chemistry, Anna University, Chennai 600 025, India, ^dGerman Wool Research Institute (DWI), Chair of Textile and Macromolecular Chemistry, University of Technology (RWTH - AAchen), Pauwelsstrasse 8.52074 AAchen, Germany, and ^eX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

Correspondence e-mail: kali@annauniv.edu, hkfun@usm.my

Key indicators

Single-crystal X-ray study T = 273 K Mean σ (C–C) = 0.002 Å Disorder in main residue R factor = 0.062 wR factor = 0.212 Data-to-parameter ratio = 33.0

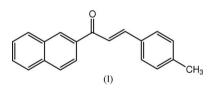
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. 3-(4-Methylphenyl)-1-(2-naphthyl)prop-2-en-1-one

The title molecule, $C_{20}H_{16}O$, has an *s*-*cis* conformation for the ketone system. The dihedral angle between the two aromatic ring systems is 50.3 (1)°. Centrosymmetrically related molecules are linked *via* weak $C-H\cdots\pi$ interactions into a chain along [110].

Received 12 October 2005 Accepted 21 October 2005 Online 27 October 2005

Comment

Chalcone is a unique template that is associated with several biological activities. Chalcone and its analogues are relatively easily available, not only by isolation from natural products but also by classical and combinatorial synthesis. The cytotoxic, anticancer, antiviral, antiprotozoal, insecticidal, chemopreventative, mutagenic and enzyme-inhibitory properties of a number of chalcones have been reviewed by Dimmock et al. (1999) and Go et al. (2005). The antifungal and antibacterial activities of these compounds have also been reviewed (Opletalova & Sedivy, 1999; Opletalova, 2000). Chalcones and their analogues are also used as potential therapeutic agents in diseases of the cardiovascular system. The stabilizing action of chalcones on the vascular wall, vasodilating and antioxidative activity have been reported (Opletalova et al., 2003). The crystal structure determination of the title compound, (I), was undertaken as part of our study of chalcones.



The molecule of (I) (Fig. 1) exhibits an *s-cis* conformation for the ketone system, defined by the torsion angle O1-C9-C8-C7 of 14.9 (2)°. A similar result was observed for 1-(2-

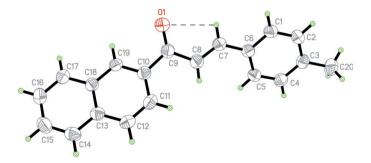
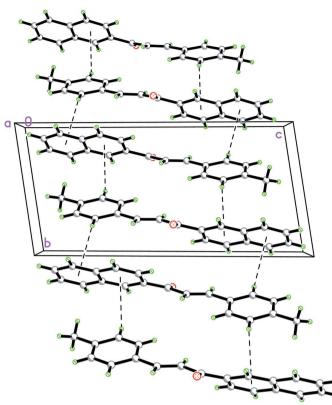


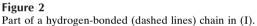
Figure 1

The structure of (I), showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. The dashed line indicates a hydrogen bond.

Printed in Great Britain - all rights reserved

© 2005 International Union of Crystallography





naphthyl)-3-(4-nitrophenyl)prop-2-en-1-one [(II); -19.4 (6)°; Shanmuga Sundara Raj et al., 1996], 3-(2-chlorophenyl)-1-(2naphthyl)prop-2-en-1-one [(III); -21.4 (3)°; Kumaran et al., 1996], 3-(4-chlorophenyl)-1-(2-naphthyl)prop-2-enone [(IV); $-20.8 (4)^{\circ}$; Shanmuga Sundara Raj *et al.*, 1997] and 1-(2-naphthalenyl)-3-(3-nitrophenyl)prop-2-en-1-one [(V): $-14.9 (3)^{\circ}$; Shanmuga Sundara Raj *et al.*, 1998]. The enone group of the molecule, containing atoms C7, C8, C9 and O1, is planar to within 0.060 (1) Å. The naphthalene ring system is planar, with atom C19 showing the largest out-of-plane displacement of 0.018 (1) Å. The mean plane through the enone group makes dihedral angles of 22.8 (1) and 28.2 $(1)^{\circ}$, respectively, with the benzene ring and naphthalene ring system. The dihedral angle between the two aromatic ring systems (C1–C6 and C10–C19) in (I) is $50.3 (1)^{\circ}$, whereas those in (II), (III), (IV) and (V) are 44.2 (1), 27.68 (5), 49.80 (8) and 2.2 (1)°, respectively.

Bond lengths and angles in (I) are comparable with those reported for (II), (III), (IV) and (V). The short H5...H8 (2.30 Å) contact causes the bond angles C5-C6-C7 $[123.42 (11)^{\circ}]$ and C6-C7-C8 $[127.41 (12)^{\circ}]$ to deviate significantly from 120°. Also, the short H8···H11 (2.32 Å) contact results in a slight widening of the C9-C10-C11 angle to 122.14 (11)°.

As seen in Fig. 2, centrosymmetrically related molecules are linked via weak C-H··· π interactions, involving the C13-C18 (centroid Cg1) and C10-C13/C18/C19 (centroid Cg2) benzene rings of the naphthalene ring system (Table 1), into a chain along [110].

Experimental

Compound (I) was obtained by Claisen-Schmidt condensation of 2acetonaphthone (1.70 g, 0.01 mol) and 4-methylbenzaldehyde (1.20 g, 0.01 mol) in ethanol (25 ml) in the presence of NaOH. The product was isolated by filtration and washed with dilute hydrochloric acid and then with water. The crude product was recrystallized from ethanol.

Crystal data

٦

$C_{20}H_{16}O$	Z = 2
$M_r = 272.33$	$D_x = 1.261 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 5.9174 (1) Å	Cell parameters from 2939
b = 7.7180(1) Å	reflections
c = 15.9606 (4) Å	$\theta = 2.6-33.4^{\circ}$
$\alpha = 81.239 \ (2)^{\circ}$	$\mu = 0.08 \text{ mm}^{-1}$
$\beta = 84.617 \ (2)^{\circ}$	T = 273 (2) K
$\gamma = 88.716 \ (1)^{\circ}$	Needle, colourless
$V = 717.22 (2) \text{ Å}^3$	$0.67 \times 0.13 \times 0.07 \; \mathrm{mm}$

6301 independent reflections

 $R_{\rm int} = 0.038$ $\theta_{\rm max} = 35.1^{\circ}$

 $h = -9 \rightarrow 9$

 $k = -12 \rightarrow 12$ $l = -25 \rightarrow 25$

3154 reflections with $I > 2\sigma(I)$

Data collection

Bruker SMART APEX2 CCD areadetector diffractometer ω scans Absorption correction: multi-scan (SADABS; Bruker, 2005) $T_{\min} = 0.948, T_{\max} = 0.995$ 21324 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.1034P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.062$	+ 0.0041P]
$wR(F^2) = 0.212$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} = 0.001$
6301 reflections	$\Delta \rho_{\rm max} = 0.24 \text{ e} \text{ Å}^{-3}$
191 parameters	$\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

Table 1	
Hydrogen-bond geometry (Å	°)

$D - H \cdots A$	<i>D</i> -Н	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C7 - H7 \cdots O1$ $C1 - H1 \cdots Cg1^{i}$ $C4 - H4 \cdots Cg2^{ii}$	0.93	2.49	2.8126 (16)	101
	0.93	2.97	3.6189 (14)	128
	0.93	2.94	3.6189 (14)	131

Symmetry codes: (i) -x, -y, -z + 1; (ii) -x + 1, -y + 1, -z + 1. Notes: Cg1 and Cg2 are the centroids of the C13-C18 and C10-C13/C18/C19 benzene rings, respectively.

H atoms were placed in calculated positions and constrained to ride on their parent atoms, with C-H = 0.93-0.96 Å and $U_{iso}(H)$ = 1.2–1.5 $U_{eq}(C)$. The methyl group was found to be disordered over two positions rotated from each other by 60° . The occupation factors were fixed at 0.5.

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

HKF thanks the Malaysian Government and Universiti Sains Malaysia for the Scientific Advancement Grant Allocation (SAGA) grant No. 304/PFIZIK/635003/A118.

References

- Bruker (2005). *APEX2* (Version 1.27), *SAINT* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Dimmock, J. R., Elias, D. W., Beazely, M. A. & Kandepu, N. (1999). Curr Med Chem. 6, 1125–1149.
- Go, M. L., Wu, X. & Liu, X. L. (2005). Curr Med Chem. 12, 481-499.
- Kumaran, D., Eswaramoorthy, S., Ponnuswamy, M. N., Raju, K. S. & Nanjundan, S. (1996). *Acta Cryst.* C**52**, 2543–2545.
- Opletalova, V. (2000). Ceska Slov. Farm. 49, 278-284.

- Opletalova, V., Jahodar, L., Jun, D. & Opletal, L. (2003). Ceska Slov. Farm. 52, 12–19.
- Opletalova, V. & Sedivy, D. (1999). Ceska Slov. Farm. 48, 252-255.
- Shanmuga Sundara Raj, S., Ponnuswamy, M. N., Shanmugam, G. & Nanjundan, S. (1996). Acta Cryst. C52, 3145–3146.
- Shanmuga Sundara Raj, S., Ponnuswamy, M. N., Shanmugam, G. & Nanjundan, S. (1997). Acta Cryst. C53, 917–918.
- Shanmuga Sundara Raj, S., Ponnuswamy, M. N., Shanmugam, G. & Nanjundan, S. (1998). Acta Cryst. C54, 541–542.
- Sheldrick, G. M. (1998). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.